Science (086) Class X

Sample Question Paper 2022-23

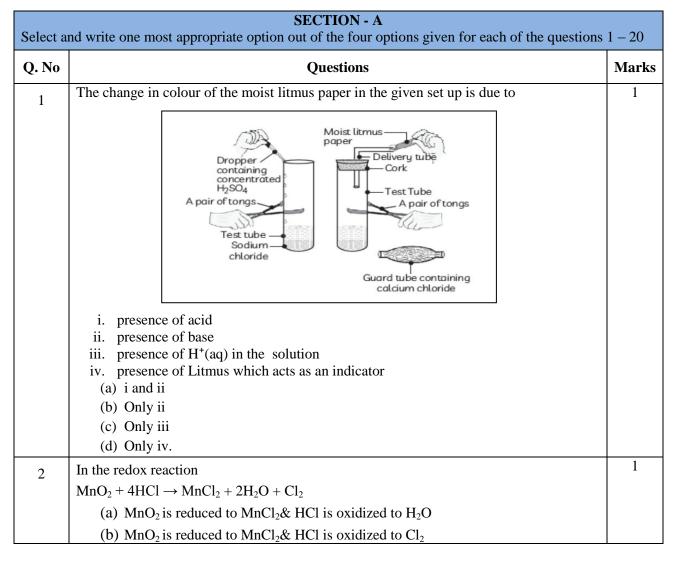
Max. Marks: 80

Time Allowed: 3 hours

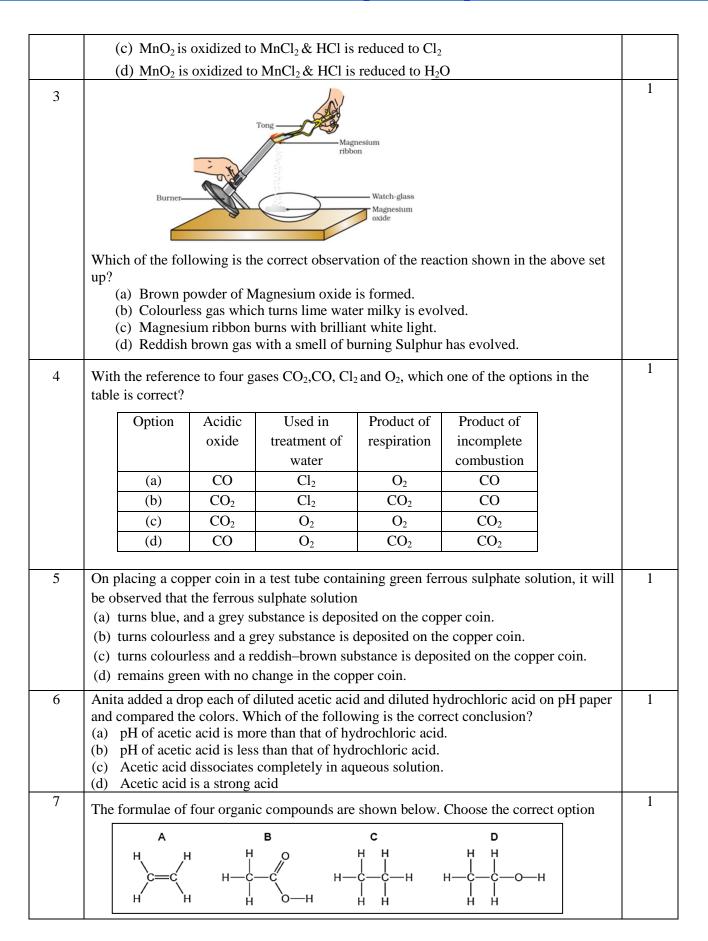
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General Instructions:

- i. This question paper consists of 39 questions in 5 sections.
- *ii.* All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. Section A consists of 20 objective type questions carrying 1 mark each.
- *iv.* Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should in the range of 30 to 50 words.
- v. Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should in the range of 50 to 80 words
- vi. Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

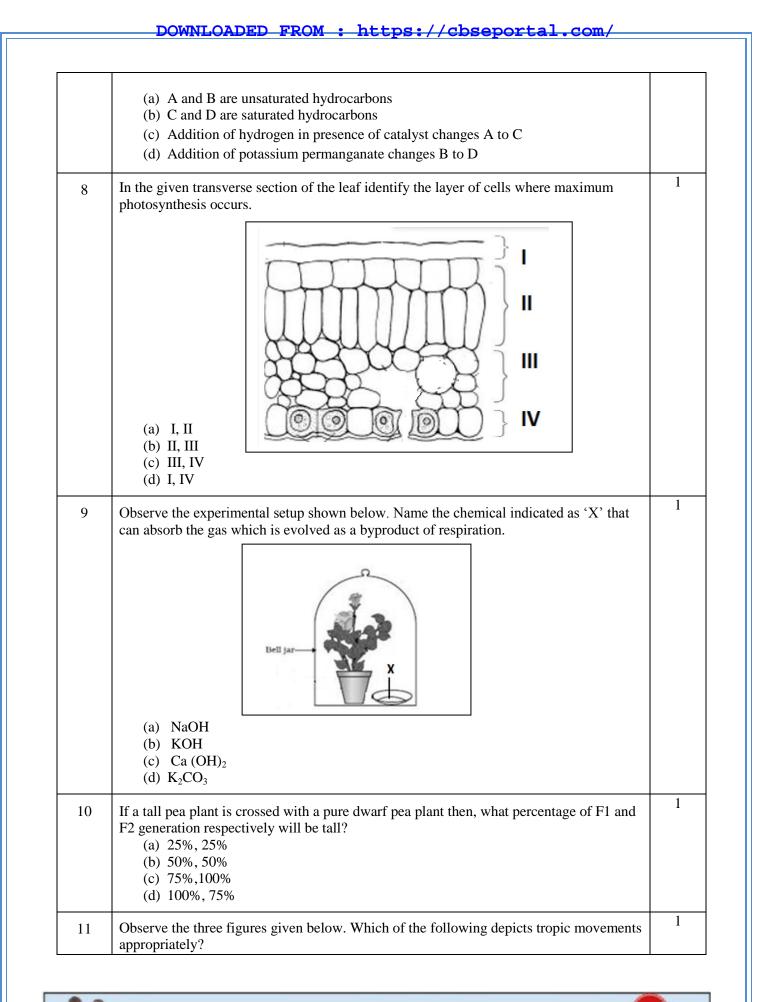






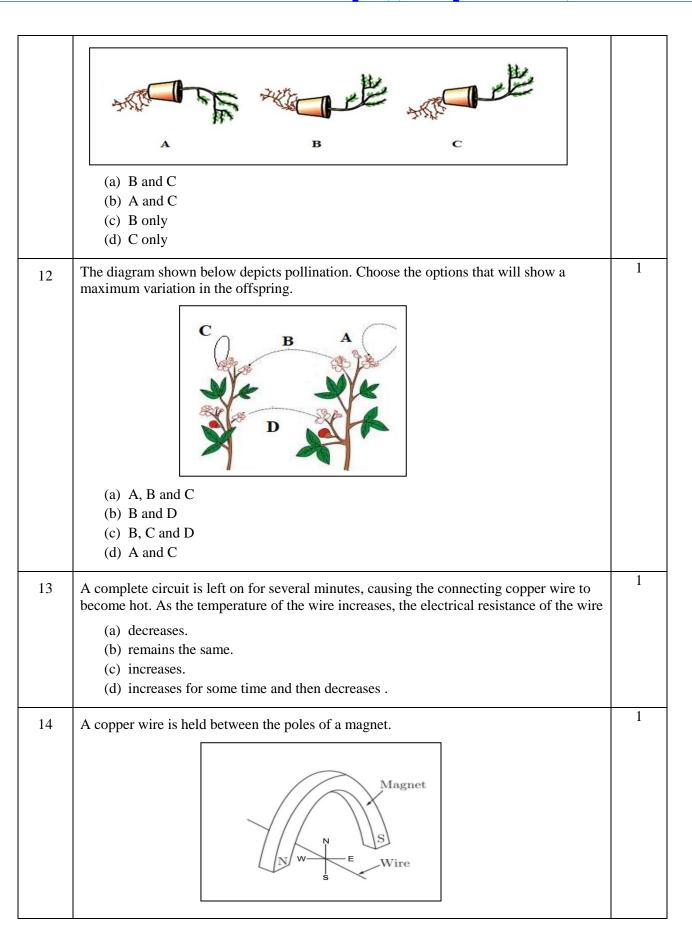




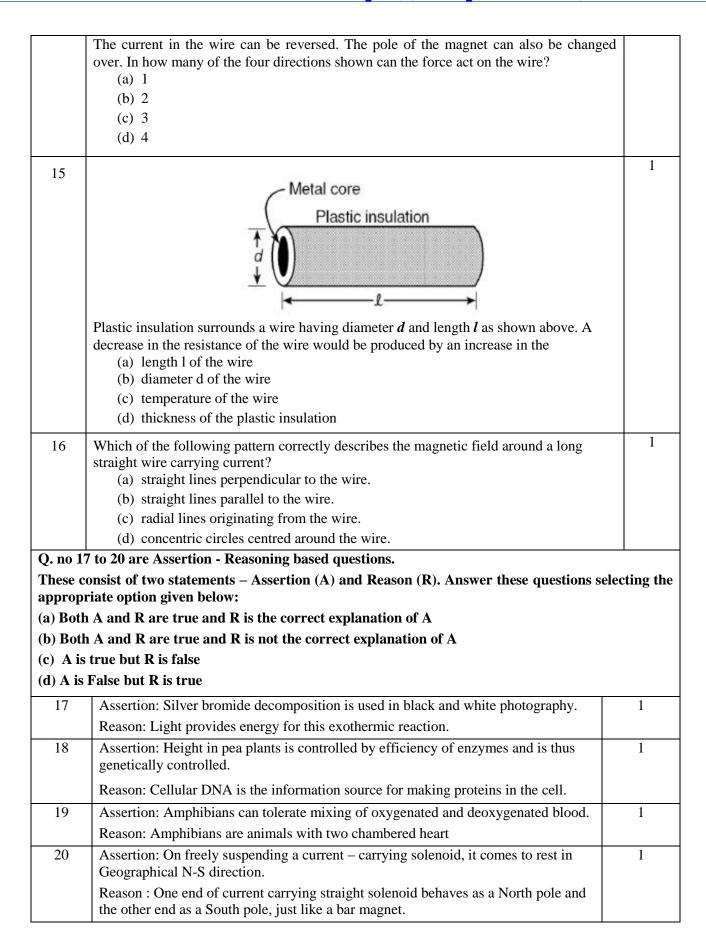




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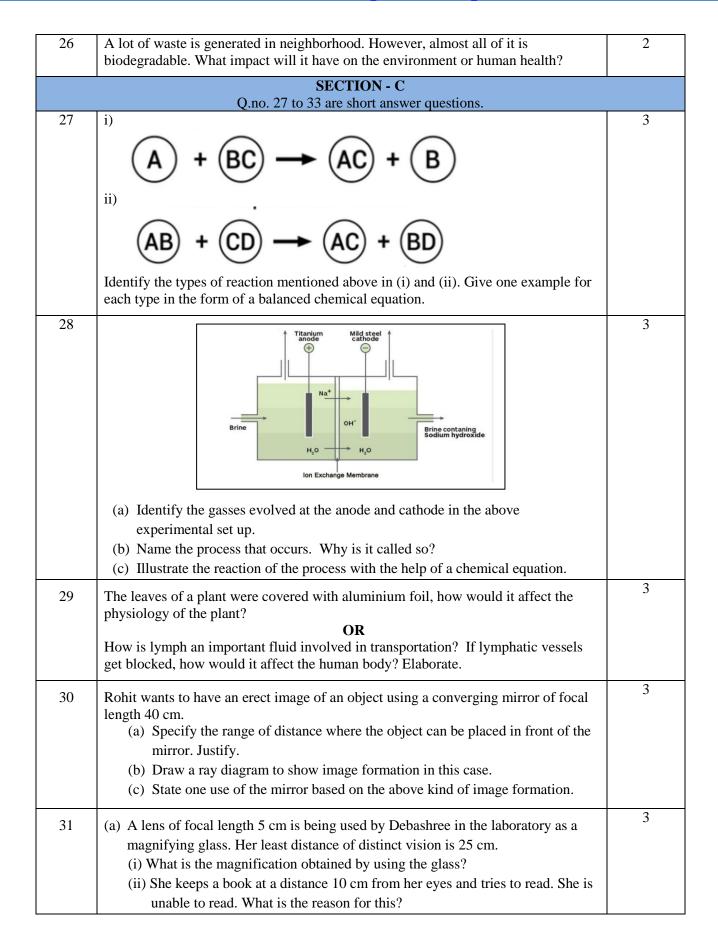


			SECTION – B	westions	
21	Q. no. 21 to 26 are very short answer questions. A clear solution of slaked lime is made by dissolving Ca(OH) ₂ in an excess of water. This solution is left exposed to air. The solution slowly goes milky as a faint white precipitate forms. Explain why a faint white precipitate forms, support your response with the help of a chemical equation. OR Keerti added dilute Hydrochloric acid to four metals and recorded her observations as shown in the table given below:				2
		Metal	Gas Evolved		
		Copper	Yes		
		Iron	Yes		
		Magnesium	No		
		Zinc	Yes		
	Select the correct involved.	observation(s) and	d give chemical equation	on(s) of the reaction	
22	How is the mode Give four example		g of the heart different	from reflex actions?	2
23	Patients whose ga Why?	allbladder are remo	oved are recommended	to eat less oily food.	2
24	Name the substances other than water, that are reabsorbed during urine formation. What are the two parameters that decide the amount of water that is reabsorbed in the kidney?			2	
25	ra	diation from the Sun		llow light ue light	2
	State the phenomena observed in the above diagram. Explain with reference to the diagram, which of the two lights mentioned above will have the higher wavelength?				
	OR				
	How will you use two identical prisms so that a narrow beam of white lightincident on one prism emerges out of the second prism as white light? Draw the diagram.				



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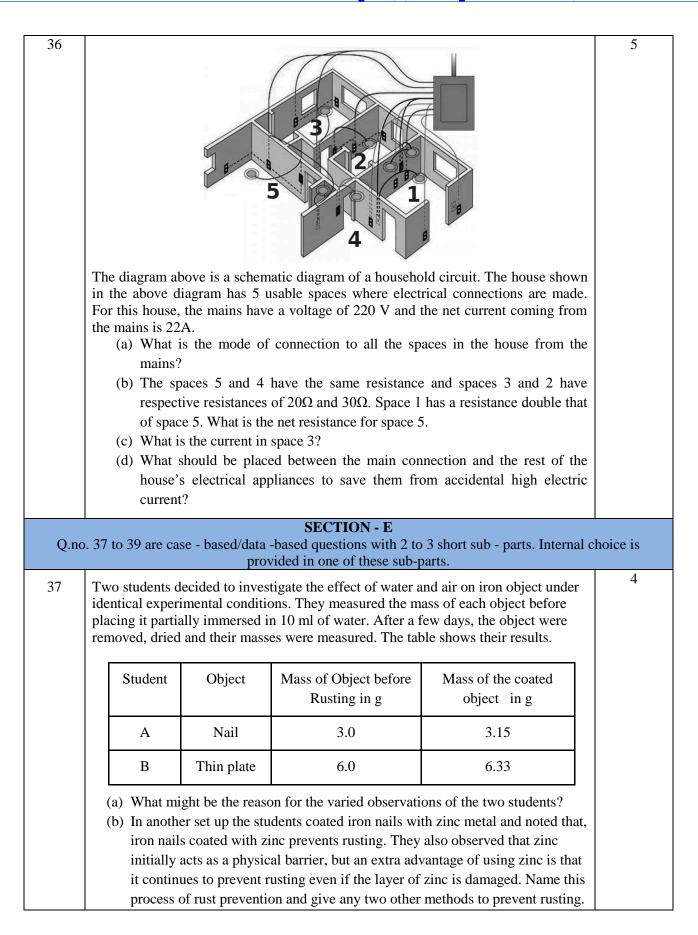


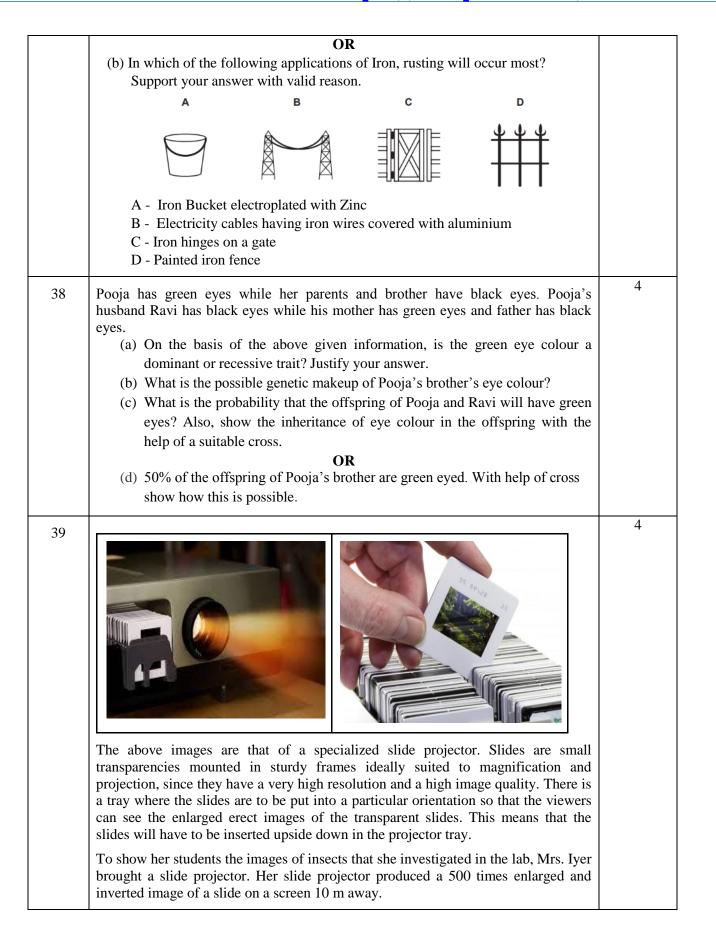


	(b) Ravi kept a book at a distance of 10 cm from the eyes of his friend Hari. Hari is not able to read anything written in the book. Give reasons for this?	
32	 A student fixes a white sheet of paper on a drawing board. He places a bar magnet in the centre and sprinkles some iron filings uniformly around the bar magnet. Then he taps gently and observes that iron filings arrange themselves in a certain pattern. (a) Why do iron filings arrange themselves in a particular pattern? (b) Which physical quantity is indicated by the pattern of field lines around the bar magnet? (c) State any two properties of magnetic field lines. OR A compass needle is placed near a current carrying wire. State your observations for the following cases and give reasons for the same in each case- (a) Magnitude of electric current in wire is increased. (b) The compass needle is displaced away from the wire. 	3
33	Why is damage to the ozone layer a cause for concern? What are its causes and what steps are being taken to limit this damage?	3
	SECTION - D	
	Q.no. 34 to 36 are Long answer questions.	E
34	 Shristi heated Ethanol with a compound A in presence of a few drops of concentrated sulphuric acid and observed a sweet smelling compound B is formed. When B is treated with sodium hydroxide it gives back Ethanol and a compound C. (a) Identify A and C (b) Give one use each of compounds A and B. (c) Write the chemical reactions involved and name the reactions. 	5
	OR	
	 (a) What is the role of concentrated Sulphuric acid when it is heated with Ethanol at 443 K. Give the reaction involved. (b) Reshu by mistake forgot to label the two test tubes containing Ethanol and Ethanoic acid. Suggest an experiment to identify the substances correctly? Illustrate the reactions with the help of chemical equations 	
35	(a) Why is it not possible to reconstruct the whole organism from a fragment in complex multicellular organisms?(b) Sexual maturation of reproductive tissues and organs are necessary link for reproduction. Elucidate.	5
	OR	
	(a) How are variations useful for species if there is drastic alteration in the niches?(b) Explain how the uterus and placenta provide necessary conditions for proper growth and development of the embryo after implantation?	











(c)When a slide is placed 15 cm behind the lens in the projector, an image is formed 3 m in front of the lens. If the focal length of the lens is 14 cm, draw a ray diagram to show image formation. (not to scale)
OR
be placed from the slide projector's lens so that the slide is in focus?
is placed upside down 21 cm from the lens. How far away should the screen
(c) A slide projector has a convex lens with a focal length of 20 cm. The slide
with one reason state what will be the sign for $\frac{v}{u}$ in the given case?
(b) If v is the symbol used for image distance and u for object distance then
must the slide projector have?
(a) Based on the text and data given in the above paragraph, what kind of lens



